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energy—the energy of hot water that exists naturally underground—as the heat source. If a supply of hot water at $140\text{ }^{\circ}\text{C}$ is discovered at a location where the environmental temperature is $20\text{ }^{\circ}\text{C}$, determine the maximum thermal efficiency a geothermal

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power plant built at that location
can have.

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the universe where observations
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Having said that in Class 11,
students will come across the topic
in Chapter 6 of the chemistry

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textbook and they have to examine a lot of concepts related to thermodynamics. They will be learning about the interchange between heat and work, laws of thermodynamics, calorimetry, reaction enthalpy, etc.

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and Chemical Thermodynamics

Wyatt Tenhaeff Milo Koretsky

Department of Chemical

Engineering Oregon State

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(a) The Clausius-Clapeyron equation: $dP_{i, \text{sat}} / P_{i, \text{sat}} = h_{i, \text{vap}} / RT^2$ or $\ln P_{i, \text{sat}} / 101 \text{ kPa} = (h_{i, \text{vap}} / R) (1/T - 1/373 \text{ [K]})$ so $P_{i, \text{sat}} = 101 \dots$

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...

Second law of thermodynamics
helps in defining the process and
also the direction of the processes.

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It tells about the possibility of a process or the restriction of a process. It helps in analyzing the performance of processes, cycles, devices etc.

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highly specialized texts and monographs that cover only parts of macromolecular science. The text intends to survey the whole field of macromolecular science. Its organization results from the following considerations. The chemical structure of

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macromolecular compounds should be independent of the method of synthesis, at least in the ideal case. Part I is thus concerned with the chemical and physical structure of polymers. Properties depend on structure. Solution properties are thus discussed in

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Part II, solid state properties in Part III. There are other reasons for discussing properties before synthesis: For example, it is difficult to understand equilibrium polymerization without knowledge of solution thermodynamics, the gel effect without knowledge of

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and topics that appear prominently in current polymer science journals. Polymer Chemistry, Third Edition offers a logical presentation of topics that can be scaled to meet the needs of introductory as well as more advanced courses in chemistry,

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This text is the published version of many of the talks presented at two symposiums held as part of the Southeast Regional Meeting of the American Chemical Society

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(SERMACS) in Knoxville, TN in October, 1999. The Symposia, entitled Solution Thermodynamics of Polymers and Computational Polymer Science and Nanotechnology, provided outlets to present and discuss problems of current interest to polymer

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Scientists. It was, thus, decided to publish both proceedings in a single volume. The first part of this collection contains printed versions of six of the ten talks presented at the Symposium on Solution Thermodynamics of Polymers organized by Yuri B.

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Melnichenko and W. Alexander
Van Hook. The two sessions,
further described below,
stimulated interesting and
provocative discussions. Although
not every author chose to
contribute to the proceedings
volume, the papers that are

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included faithfully represent the scope and quality of the symposium. The remaining two sections are based on the symposium on Computational Polymer Science and Nanotechnology organized by Mark D. Dadmun, Bobby G. Sumpter, and

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Don W. Noid. A diverse and distinguished group of polymer and materials scientists, biochemists, chemists and physicists met to discuss recent research in the broad field of computational polymer science and nanotechnology. The two-day oral

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session was also complemented by a number of poster presentations.

The first article of this section is on the important subject of polymer blends. M. D.

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