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How to Master

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reviews: Kaplan  
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C: When  $n = 3$ ,  $l$   
 $= 0, 1, \text{ or } 2$ . The

highest value  
for  $l$  in this  
case is  $2$ , which  
corresponds to  
the  $d$  subshell.

Although the  $3d$   
block appears to  
be part of the  
fourth period,

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it still has the principal quantum number  $n = 3$ . In general, the subshells within an energy shell increase in energy as follows:  $s < p < d < f$  (although there is no  $3f$  subshell).

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I condensed

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EVERYTHING from

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books into 89

pages of notes.

(Except CARS) I

repeat: I sat

down for 200

hours and took

notes on EVERY

SINGLE equation,

important

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concept,

pathway, etc. I scored 514 on the actual MCAT, which I took in August 2018.

Hopefully this helps somebody! Happy studying!

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